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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.029 wR factor = 0.078 Data-to-parameter ratio = 14.8

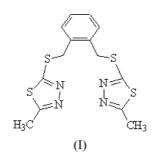
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2,2'-[1,2-Phenylenebis(methylenethio)]bis(5-methyl-1,3,4-thiadiazole)

The title compound (PBT), $C_{14}H_{14}N_4S_4$, possesses C_2 symmetry, with the twofold axis bisecting the central benzene ring. The two terminal 2-(5-methyl-1,3,4-thiodiazolyl)thio groups adopt a *trans* conformation with respect to the benzene moiety, and the dihedral angle between the two thiodiazole planes is 61.3 (2)°, while that between the thiodiazole and benzene planes is 79.4 (5)°.

Comment

In current coordination and supramolecular chemistry, the rational design of coordination architectures represents one of the most exciting and rapidly developing fields, owing to their potential as functional materials (Braga et al., 1998). The design of ligands is important in adjusting the coordination framework, and the alternation of the coordinating donor may lead to the formation of unique frameworks with tailored properties and functions. Heteroaryl thioether ligands have shown interesting coordination chemistry with metal ions, but they are mainly restricted to pyridine derivatives (Hartshorn et al., 1998), and a small number of examples with other heteroaryl thioether ligands have been reported (Hong et al., 2000; Yang et al., 1997). Herein we report the synthesis and structure of a new ligand, (I) (PBT), which contains two 2-(5methyl-1,3,4-thiadiazolyl)thio groups separated by an o-xylenediyl spacer.



As shown in Fig. 1, the molecule of PBT possesses C_2 symmetry, with the twofold axis bisecting the central benzene ring. The two terminal 2-(5-methyl-1,3,4-thiodiazolyl)thio groups adopt a *trans* conformation with respect to the benzene moiety, and the dihedral angle between the two thiodiazole planes is 61.3 (2)°. The pseudo-torsion angle of the two C–S bonds (S2–C4···C4A–S2A) is 139.9 (4)°, forcing the two thiodiazole rings to point in opposite directions and reducing the steric repulsion between them. The dihedral angles between the two terminal groups and the central benzene ring are both 79.4 (5)°.

Because of the π - π conjugation of S2 with the thiodiazole ring, the bond distance C3-S2 [1.746 (2) Å] is obviously

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shorter than C4–S2 [1.825 (2) Å]. A similar effect has been found in 2,2'-[1,4-phenylenebis(methylenethio)]bisthiazole (Zhang *et al.*, 2003).

Experimental

The title compound was synthesized as follows: 5-methyl-2-sulfanyl-1,3,4-thiodiazole (1.32 g, 10 mmol) was added to a stirred solution of KOH (0.56 g, 10 mmol) in ethanol (20 ml). The mixture was warmed to reflux, then 1,2-dibromomethylbenzene (1.32 g, 5 mmol) in tetrahydrofuran (5 ml) was added dropwise and the mixture was refluxed for about 6 h. The precipitate was filtered off and washed with water and ethanol, giving a fine white powder in 80% yield. m.p.: 387– 388 K; IR (KBr): 2944 (*m*), 2854 (*w*), 1635 (*w*), 1488 (*s*), 1437 (*vs*), 1379 (*s*), 1365 (*vs*), 1188 (*vs*), 1070 (*vs*), 1038 (*vs*), 789 (*s*), 748 (*m*); ¹H NMR (CDCl₃): δ 2.72 (*s*, 6H, CH₃), 4.70 (*s*, 4H, CH₂), 7.35 (*m*, 4H, Ar-H); analysis calculated for C₁₄H₁₄N₄S₄: C 45.84, H 3.82, N 15.28%; found: C 45.56, H 3.68, N 14.92%. Colorless single crystals were obtained by recrystallized from chloroform and methanol.

Crystal data

$C_{14}H_{14}N_4S_4$	$D_x = 1.429 \text{ Mg m}^{-3}$
$M_r = 366.53$	Mo $K\alpha$ radiation
Monoclinic, $C2/c$	Cell parameters from 774
a = 18.100 (10) Å	reflections
b = 9.239(5) Å	$\theta = 2.5 - 26.3^{\circ}$
c = 11.216 (6) Å	$\mu = 0.56 \text{ mm}^{-1}$
$\beta = 114.764 \ (8)^{\circ}$	T = 293 (2) K
$V = 1703.3 (16) \text{ Å}^3$	Block, colorless
Z = 4	$0.30\times0.25\times0.20~\text{mm}$

Data collection

Bruker SMART CCD area-detector diffractometer	1497 independent reflections 1242 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\rm int} = 0.038$
Absorption correction: multi-scan	$\theta_{\rm max} = 25.0^{\circ}$
(SADABS; Sheldrick, 1996;	$h = -21 \rightarrow 19$
Blessing, 1995)	$k = -11 \rightarrow 10$
$T_{\min} = 0.851, T_{\max} = 0.897$	$l = -13 \rightarrow 13$
2152 measured reflections	

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0345P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.029$	+ 1.2785P]
$wR(F^2) = 0.078$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.08	$(\Delta/\sigma)_{\rm max} = 0.002$
1497 reflections	$\Delta \rho_{\rm max} = 0.16 \ {\rm e} \ {\rm \AA}^{-3}$
101 parameters	$\Delta \rho_{\rm min} = -0.27 \text{ e} \text{ \AA}^{-3}$
H-atom parameters constrained	

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve

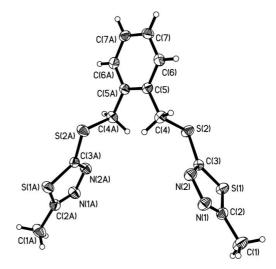


Figure 1

View of the title compound, showing the atom-numbering scheme and displacement elliposids at the 30% probability level.

structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1998); software used to prepare material for publication: *SHELXTL*.

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References

- Blessing, R. H. (1995). Acta Cryst. A51, 33-38.
- Braga, D., Grepioni, F. & Desiraju, G. R. (1998). *Chem. Rev.* 98, 1375–1405.Bruker (1998). *SMART*, *SAINT* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Hartshorn, C. M. & Steel, P. J. (1998). J. Chem. Soc. Dalton Trans. pp. 3935– 3940.
- Hong, M. C., Su, W. P., Cao, R., Fujita, M. & Lu, J. X. (2000). Chem. Eur. J. 6, 427–431.
- Yang, C., Chen, X. M., Lu, X. Q., Zhou, Q. H. & Yang, Y. S. (1997). Chem. Commun. pp. 2041–2042.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Zhang, W., Liu, H. M., Li, C. B. & Zhang, W. Q. (2003). Acta Cryst. E**59**, o26–o27.